ORIGINAL ARTICLE



Two New Cytotoxic Quinone Type Compounds from the Halotolerant Fungus *Aspergillus variecolor*

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Abstract Two new quinone type compounds, variecolorquinones A (1) and B (2) together with eleven known related compounds $3\sim13$ have been isolated from the metabolites produced by the halotolerant fungal strain *Aspergillus variecolor* B-17. The structures of 1 and 2 were determined by spectroscopic methods. 1 exhibited selective cytotoxicity against A-549 cells with the IC₅₀ values of 3.0 μ M. 2 showed cytotoxicity against HL60 and P388 cells with the IC₅₀ values of 1.3 and 3.7 μ M, respectively.

Keywords variecolorquinone, cytotoxic, metabolite, quinone, halotolerant fungi

Introduction

It demonstrated that halotolerant marine fungal species have evolved unique metabolic mechanisms that are responsive to salt concentrations and marine-derived fungal metabolite production could have implications for drug discovery [1]. Besides marine samples, halotolerant fungal species had been isolated from other hypersaline environment such as salterns, salines and dried fish. But, to the best of our knowledge, there were only a few reports on secondary metabolites of halotolerant fungus in the literature $[2\sim4]$. In order to screen "talented strains" [5] to find structural novel and bioactive secondary metabolites,

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we continue to isolate and identify halotolerant microbes from both marine [$6\sim13$] and other salt environments. As a result, a halotolerant fungal strain, *Aspergillus variecolor* B-17, was isolated from the sediments collected in Jilantai salt field, Alashan, Inner Mongolia, China.

The crude extract of the mycelia of *A. variecolor* B-17 showed significant cytotoxicity against the mouse *cdc2* mutant cell line (tsFT210) and was subjected to flash column chromatography over silica gel, RP-18 and HPLC separation to afford two new quinone type compounds variecolorquinones A (1) and B (2), together with eleven known compounds, (2*S*)-2,3-dihydroxypropyl 1,6,8-trihydroxy-3-methyl-9,10-dioxoanthracene-2-carboxylate (3) [14], emodin (4) [15], physcion (5) [15], questin (6) [15], questinol (7) [15], catenarin (8) [15], erythroglaucin (9) [16], rubrocristin (10) [16], fallacinol (11) [15], eurotinone (12) [17] and 2-methyleurotinone (13) [17] (Fig. 1), by a bioassay-guided isolation procedure.

Materials and Methods

Strain

The working strain *Aspergillus variecolor* B-17 was isolated from the sediments collected in Jilantai salt field, Alashan, Inner Mongolia, China. It was identified by Prof. Li Tian, the First Institute of Oceanography, SOA, Qingdao, China. Working strain was prepared on Potato Dextrose agar slants containing 10% NaCl and stored at 4°C.

Fermentation

The fungus was grown under static conditions at 30°C for 28 days in thirty 1000-ml conical flasks containing the liquid medium (300 ml/flask) composed of (g/liter): glucose

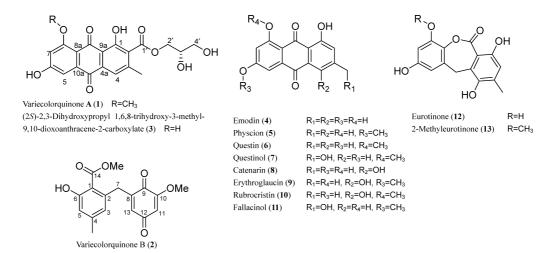


Fig. 1 Structures of variecolorquinones A (1) and B (2), eleven known compounds 3~13.

(10), maltose (20), mannitol (20), malt extract (3.0), monosodium glutamate (10), NaCl (120), NH₄Cl (10), MgSO₄·7H₂O (5.0) and KCl (5.0) after adjusting its pH to 7.0.

Extraction and Isolation

The fermented whole broth (9.0 liters) was filtered through cheese cloth to separate into supernatant and mycelia. The mycelia was extracted three times with Me₂CO. The Me₂CO solution was concentrated under reduced pressure to afford crude extract (7.8 g), showing strong cytotoxicity against the mouse cdc2 mutant cell line (tsFT210, IC₅₀ $68 \,\mu\text{g/ml}$) and antioxidation against 1,1-diphenyl-2pierylhydrazyl (DPPH, IC $_{50}$ 98 $\mu g/ml$). The crude extract was separated into 4 fractions on a silica gel column using a step gradient elution of CHCl₃-MeOH. The fraction 1 (1.9 g) was separated into 5 subfractions on a silica gel column using a step gradient elution of petroleum ether: Me₂CO. Subfraction 1-4 (73 mg) was crystallized from CHCl₃/MeOH (v/v 9:1) to yield **5** (39 mg). **9** (11 mg) was isolated from the mother liquid of subfraction 1-4 by Sephadex LH-20 column (v/v 2:1 CHCl₃/MeOH). The fraction 3 (0.8 g) was subjected to column chromatography over silica gel (v/v 93:7 CHCl₃/MeOH) to afford 16 subfrations. Subfraction 3-2 (72 mg) was crystallized from CHCl₃/MeOH (v/v 1:1) to yield 8 (59 mg). The mother liquid of subfraction 3-2, togather with the Fr. 3-3 and 3-4 were combined and subjected to HPLC separation (gradient elution of $70\sim100\%$ MeOH) to yield 4 (5.0 mg), 6 (7.0 mg), 7 (9.0 mg), 10 (7.0 mg) and 11 (6.0 mg). The Fr. 3-7 (59 mg) was further separated by HPLC to yield 2 (31 mg). Subfractions 3-14 (29 mg) and 3-15 (83 mg) were further separated by HPLC to yield 1 (15 mg) and 3 (45 mg), respectively. Subfractions 3-5 and 3-6 were combined and subjected to HPLC to yield 12 (31 mg) and 13 (23 mg).

Physico-chemical Analyses

Optical rotations were obtained on a JASCO P-1020 digital polarimeter. UV spectra were recorded on Beckmen DU® 640 spectrophotometer. IR spectra were taken on a NICOLET NEXUS 470 spectrophotometer in KBr discs. 1 H-, 13 C-NMR and DEPT spectra and 2D-NMR were recorded on a JEOL JNM-ECP 600 spectrometer using TMS as internal standard and chemical shifts were recorded as δ values. ESI-MS was measured on a Q-TOF ULTIMA GLOBAL GAA076 LC mass spectrometer. Semipreparative HPLC was performed on a SHIMADZU LC-6AD Liquid Chromatograph with SPD-M10A vp Diode Array Detector.

Biological Assay

Cell lines were grown in RPMI-1640 supplemented with 10% FBS under a humidified atmosphere of 5.0% $\rm CO_2$ and 95% air at 37°C. Two hundred microliters of those cell suspensions at a density of 5×10^4 cell/ml was plated in 96 well microtiter plates and incubated for 24 hours at the above conditions. Then 2.0 μ l of the test compound solutions (in MeOH) at different concentrations was added to each well and further incubated for 72 hours in the same conditions. Twenty microliters of the MTT solution (5.0 mg/ml in RPMI-1640 medium) was added to each well and incubated for 4 hours. An old medium (150 μ l) containing MTT was then gently replaced by DMSO and pipetted to dissolve any formazan crystals formed. Absorbance was then determined on a SPECTRA MAX PLUS plate reader at 540 nm.

In SRB assay, 200 μ l of the cell suspensions were plated

Table 1 1 H- and 13 C-NMR data for variecolorquinone A (1) in DMSO- d_{6}

| No. | $\delta_{ m H}$ (J in Hz) | $\delta_{\scriptscriptstyle \mathbb{C}}$ | No. | $\delta_{	extsf{H}}$ (J in Hz) | $\delta_{\scriptscriptstyle \mathbb{C}}$ |
|-----|---------------------------|--|--------------------|--------------------------------|--|
| 1 | | 158.8s | 10 | | 182.2s |
| 2 | | 129.4s | 10a | | 137.1s |
| 3 | | 143.3s | 1-OH | 13.6 (s) | |
| 4 | 7.48 (s) | 119.6d | 6-OH | 11.3 (s) | |
| 4a | | 132.6s | 3-CH ₃ | 2.37 (3H, s) | 19.9q |
| 5 | 7.19 (brs) | 107.1d | 8-OCH ₃ | 3.91 (3H, s) | 56.8q |
| 6 | | 166.3s | 1′ | | 164.0s |
| 7 | 6.83 (brs) | 105.0d | 2′ | 4.38 (dd, 10.9, 3.9) | 67.3t |
| | | | | 4.19 (dd, 10.9, 6.4) | |
| 8 | | 165.2s | 3′ | 3.76 (m) | 69.7d |
| 8a | | 113.0s | 4′ | 3.43 (dd, 11.0, 5.5) | 63.0t |
| | | | | 3.38 (dd, 11.0, 6.2) | |
| 9 | | 186.5s | 3'-OH | 5.05 (br s) | |
| 9a | | 115.1s | 4'-OH | 4.75 (br s) | |

Table 2 ¹H- and ¹³C-NMR data for variecolorguinone B (2) in CDCl₃

| No. | $\delta_{ m H}$ (J in Hz) | $\delta_{	extsf{C}}$ | HMBC (H–C) | ¹ H- ¹ H COSY |
|--------|------------------------------|----------------------|------------------|-------------------------------------|
| 1 | | 109.1s | | |
| 2 | | 138.4s | | |
| 3 | 6.53 (d, 1.3) | 125.4d | 1, 2, 5 | 5 |
| 4 | | 146.1s | | |
| 5 | 6.78 (brs) | 117.7d | 1, 3, 6, 4-Me | 3 |
| 6 | | 163.6s | | |
| 7 | 4.00 (2H, d, 1.8) | 34.9 t | 1,2, 3, 8, 9, 13 | 13 |
| 8 | | 147.7s | | |
| 9 | | 181.7s | | |
| 10 | | 158.6s | | |
| 11 | 5.87 (d, 2.2) | 107.1d | 9, 10, 12, 13 | 13 |
| 12 | | 187.4s | | |
| 13 | 5.99 (dt, 1.8, 2.2) | 132.7d | 7, 9, 11 | 7, 11 |
| 14 | | 170.8s | | |
| 4-Me | 2.32 (3H, s) | 21.6q | 3, 4, 5 | |
| 10-OMe | 3.85 (3H, s) | 56.4q | 10 | |
| 14-OMe | 3.76 (3H, s) | 52.2q | 14 | |
| 6-OH | 11.29 (s) | | 1, 5, 6 | |

in 96-cell plates at a density of 2×10^5 cell/ml. Then $2.0\,\mu l$ of the test compound solutions (in MeOH) at different concentrations was added to each well and the culture was further incubated for 24 hours. Following drug exposure, the cells were fixed with 12% TCA and the cell layer was stained with 0.4% SRB. The absorbance of SRB solution was measured at 515 nm. Dose response curves were generated and the IC₅₀ values, the concentration of

compound required to inhibit cell proliferation by 50%, were calculated from the linear portion of log dose response curves.

In DPPH scavenging assay, $160 \,\mu\text{l}$ of reaction mixtures containing test samples and $40 \,\mu\text{M}$ DPPH (Sigma) dissolved in methanol were plated in 96-cell plates and kept in dark for 30 minutes. After the reaction, absorbance was then measured at 517 nm, and percent inhibition was

Fig. 2 The key HMBC and ¹H-¹H COSY correlations of 1 and 2.

calculated. IC_{50} values denote the concentration of sample required to scavenge 50% of the DPPH free radicals.

Results and Discussion

Physico-chemical Properties

1: yellow amorphous powder; $[\alpha]_{\rm D}^{25}$ –18 (c 0.03 MeOH); HRESI-MS m/z 401.0871 (calcd for ${\rm C}_{20}{\rm H}_{17}{\rm O}_{9}$, 401.0873); UV $\lambda_{\rm max}^{\rm MeOH}$ (log ε) nm 223 (4.1), 251 (3.8), 286 (3.9), 440 (3.5); IR $v_{\rm max}$ cm⁻¹ (KBr) 3425, 2941, 1718, 1672, 1635, 1598, 1574, 1436, 1343, 1245, 1203, 1146, 1114, 1072, 1047, 962, 843. $^{1}{\rm H}$ (DMSO- d_{6} , 600 MHz) and $^{13}{\rm C}$ (DMSO- d_{6} , 150 MHz) see Table 1.

2: yellow amorphous powder; HRESI-MS m/z 315.0880 (calcd for $C_{17}H_{15}O_6$, 315.0869); UV $\lambda_{\rm max}^{\rm MeOH}$ (log ε) nm 197 (4.2), 242 (4.1), 297 (3.9), 352 (3.7); IR $\nu_{\rm max}$ cm⁻¹ (KBr) 3068, 1652, 1602, 1445, 1326, 1238, 1177. ^{1}H (CDCl₃, 600 MHz) and ^{13}C (CDCl₃, 150 MHz) see Table 2.

Structure Determination

1 was obtained as a yellow amorphous powder. Its molecular formula was determined as C₂₀H₁₈O₉ based on HRESI-MS at m/z 401.0871 [M-H]⁻ (calcd 401.0873). The diagnostic IR peaks were observed for hydroxyl and carbonyl at 3425 and 1718 cm⁻¹, respectively. The UV spectra absorptions at λ_{max} (log ε) 223 (4.1), 251 (3.8), and 286 (3.9), 440 (3.5) suggested the presence of emodin skeleton in 1 [15]. Analysis of the 1D NMR spectra of 1 displayed three carbonyls, nine sp^2 quarternary carbons, three sp^2 methines, one oxygenated methine, two oxygenated methylenes, one methoxyl and one methyl (Table 1). Except for a methoxyl signal ($\delta_{\rm H}$ 3.91 and $\delta_{\rm C}$ 56.8) instead of hydroxyl signal (δ 12.72), the ¹H- and ¹³C-NMR spectra of 1 were similar to those of 3 [14], suggesting their structures were very similar. Further comparison of the ¹³C-NMR spectra of 1 with those of 3 revealed -2.0, -3.1, -1.3, -4.9, +3.9 and +2.0 ppm of chemical shift effects for C-5, C-7, C-8, C-9, C-8a and C-10a, respectively, showing that 1 is 8-O-methyl derivative of 3. This conclusion was further confirmed by key ¹H with ¹³C long range correlations of –OCH₃ with C-8, H-5 with C-7 and C-8a, H-7 with C-5 and C-8a. The absolute configuration of C-2' was assigned as S by comparing $[\alpha]_D$ (-18°) with those of 3 ($[\alpha]_D$ -23°) [14]. Thus, the structure of 1 was elucidated as (2S)-2,3-dihydroxypropyl-1,6-dihydroxy-8-methoxy-3-methyl-9,10-dioxoanthracene-2-carboxylate.

2 was obtained as yellow amorphous powder. Its molecular formula was determined as C₁₇H₁₆O₆ based on HRESI-MS at m/z 315.0880 [M-H]⁻ (calcd 315.0869). Its ¹H spectral data revealed the signals of a hydrogen-bonded hydroxyl (1-OH, $\delta_{\rm H}$ 11.29), two pairs of meta-coupled aromatic protons ($\delta_{\rm H}$ 6.53 and 6.78; 5.87 and 5.99), two methoxyls ($\delta_{\rm H}$ 3.76 and 3.85), an sp^3 methyl ($\delta_{\rm H}$ 2.32) and an sp^3 methylene ($\delta_{\rm H}$ 4.00). The ¹³C-NMR spectra of 2 contained three carbonyls, six sp^2 quaternary carbons including two oxygenated ones, four sp^2 methines, an sp^3 methylene, two methoxyls and a methyl (Table 1). According to 2D NMR spectrum, these protons and carbons were linked into two structural moieties, i.e. methyl 2-substituted-4-methyl-6-hydroxybenzoate and 2substituted-6-methoxybenzoquinone. These two moieties were further connected together via a methylene which was confirmed by HMBC correlations between H-7 and C-1, C-2, C-3, C-8, C-9 and C-13, and between H-13 and C-7. These data are sufficient to assign the structure of 2 as methyl 2-hydroxy-6-[(5-methoxy-3,6-dioxocyclohexa-1,4dienyl)methyl]-4-methylbenzoate (Fig. 2).

Cytotoxity and Radical Scavenging Activity

The cytotoxicities of new compounds 1 and 2 were assayed *in vitro* against the P388 and HL-60 cell lines by the MTT method [18], and BEL-7402 and A-549 cell lines by the SRB method [19]. The IC₅₀ values of 1 and 2 against P388, HL-60, BEL-7402 and A-549 cells were 266, 309, 114, 3.0 and 3.7, 1.3, 29 and 56 μ M, respectively. 1 selectively inhibited the proliferation of A-549 cell and 2 inhibited the proliferation of P388 and HL-60 cells. 1, 2, 12 and 13 were also evaluated for their radical scavenging activity against DPPH [20]. 1 showed moderate activity with IC₅₀ values of 28 μ M, while 2 was inactive (IC₅₀ >100 μ M). 12 and 13 which were strong KDR kinase inhibitors [17] exhibited strong antioxidative activity against DPPH with IC₅₀ values

of 6.0 μ M and 11 μ M, respectively (ascorbic acid as positive control, IC₅₀ 22 μ M).

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